

Evaluating the impact of lead-lithium purity on tritium transport for tritium extraction system design in fusion plants

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1. Problem

Tritium control is central to safe fusion operation. Lead-lithium (PbLi) eutectic, used as breeder and multiplier, remains poorly characterized for tritium transport, with reported Sieverts' constants spanning nearly **four orders of magnitude**. Tritium solubility in PbLi varies drastically due to: (1) **Impurities** (metallic & non-metallic) increasing solubility; (2) **Lithium depletion** from burnup or evaporation reducing solubility; (3) **Corrosion products** dynamically altering alloy composition. This uncertainty prevents accurate TES (Tritium Extraction System) sizing and regulatory assessment.

2. Why lead-lithium (PbLi)

Breeder / Coolant	PbLi	Li	LiF-BeF ₂
Density (kg m ⁻³)	~9500	505	~2000
Thermal conductivity (W m ⁻¹ K ⁻¹)	15-35	40-60	0.57-1.1
Chemical reactivity	Mild	Violent	Very low
Corrosion tendency	Moderate	Moderate	Moderate
Specific heat (kJ kg ⁻¹ K ⁻¹)	0.19	4.26	2.4
Multiplier built-in	Yes	No	Yes
Concerning activation products	²¹⁰ Po	No	¹⁰ B, ¹⁸ F, ²⁰ F
Tritium extraction practicality	Proven, continuous	Difficult / batchwise	Slow, chemistry-limited
Tritium permeation in struct. mat.	High	Intermediate (LiT)	Low (ionic TF, HF)

- ✓ PbLi offers **greater tritium production** and energy capture per unit blanket thickness.
- ⊗ However, it can suffer from low specific heat (leading to sharper temperature gradients) and severe **tritium permeation** issues in structural materials.

3. Lead-lithium Fabrication

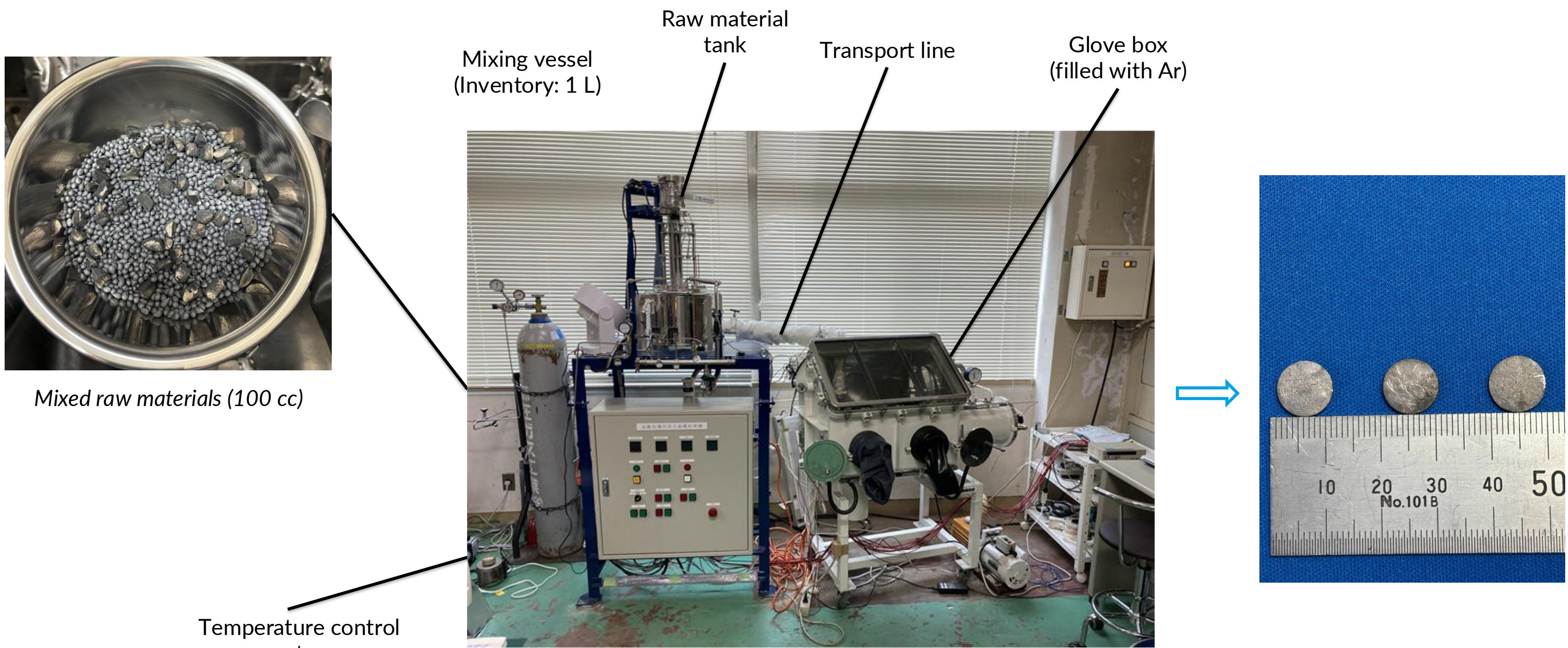


Fig 1. EX-Fusion PbLi Lab (Tokyo)

- EX-Fusion produces PbLi alloys with **commercial reactor grade purity** using low-temperature synthesis (300–340°C)
- Input metals: 99.99 wt% Pb, 99.98 wt% Li.
- Metallic impurity total < 20 ppm per batch up to **20× purer** than other commercial suppliers.

4. Solution: Experimental Set-up

INL will employ TGAP at the STAR facility to quantify tritium solubility and diffusivity in four PbLi chemistries:

- (1) High-purity eutectic PbLi
- (2) Li-depleted PbLi (Pb-5Li)
- (3) PbLi with oxide impurities
- (4) PbLi with metallic corrosion products (Fe, Cr, Mo)



Fig 3. The TGAP experiment enclosure

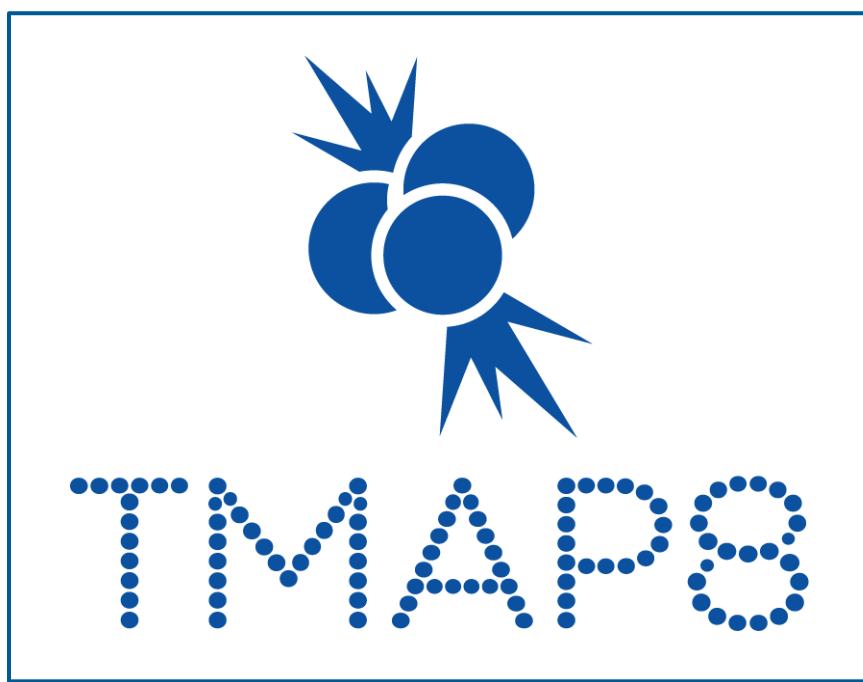
These tests will determine how PbLi chemistry influences Sieverts' constants, using tritium gas absorption and permeation kinetics. Results will form a new validated database for use in **TES models** and **blanket simulations**.



Fig 2. Tritium Gas Absorption Permeation (TGAP) Experiment

5. Solution: Simulation Validation

Data is interpreted with **TMAP8** in the MOOSE framework, **coupling** diffusion, solubility, and surface kinetics. Simulations quantify how chemistry shifts extraction efficiency and feed directly into vacuum-permeator design (PAV) for blanket systems.



6. Extraction Systems

Tritium extraction efficiency governs safety margins. In a 500 MW_{th} pilot reactor (FNSF-DCLL), extraction efficiency of **95%** corresponds to **6.2 g T/y** permeation loss, while **70%** efficiency raises losses to **38 g T/y**, potentially exceeding NRC dose limits. By quantifying how impurities and Li depletion alter tritium solubility, this project provides the data necessary to:

1. Optimize **vacuum-permeator sizing** and **membrane selection**
2. Define **impurity thresholds** for fuel-cycle safety analyses
3. Support **qualification** of EX-Fusion PbLi as a reference material for U.S. blanket R&D

7. Market Opportunity

Type of Contract	Horizons	Quantities per deal	Anticipated revenue per deal	Anticipated number of deals
National Labs & Universities	2025+	1 – 100 kg	\$1k – \$10k	~100+ between 2025-2040
Fusion Developers (experimental scale)	2028+	10 – 200 kg	\$5k – \$100k	~10+ between 2025-2040
Fusion Developers (blanket scale)	2030+	10,000 – 100,000 kg	\$1M – \$10M	3-5 between 2030-2040

High-purity PbLi remains **unavailable** commercially. EX-Fusion's process establishes a scalable route for fusion-grade breeder and coolant supply while highlighting the need for **standardized specifications** (composition, impurity thresholds, and QA).

8. References

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